

Metal Mania

Simple Models of the Material World

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Question 5

Why are some materials insulators and some conductors?
 Why do thermal and electrical conductivity go hand-in-hand?

Bonding

Gas \rightarrow Liq \rightarrow Solid (\uparrow import of interact)



- generalizes to intermolecular bonding

- directional bonding

\Leftrightarrow
 e^- pairings
 \Leftrightarrow
 band filling
 \Leftrightarrow
 insulator/semiconductor

However, all bonding types play a role

Quantum

B.C. (Box, Lattice) \rightarrow Discrete E Lvs "Quantization"

- Quantize dynamical variable (property of system)
 Ex: \vec{L} , \vec{p} , \vec{K} (periodic), \vec{S} (intrinsic) [$S = \frac{1}{2}$]

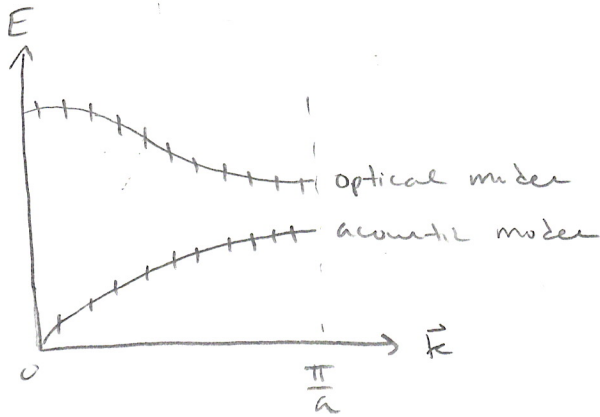
N Inputs $\rightarrow N$ States $N \rightarrow \infty \rightarrow$ Continuum

* Counter-intuitive continuous states from quantum theory

Scope: Macroscopic object
 Molecules
 Atomic
 e^-
 Nucleonic

Restrict to Crystals
 \downarrow
 Periodicity

Atoms



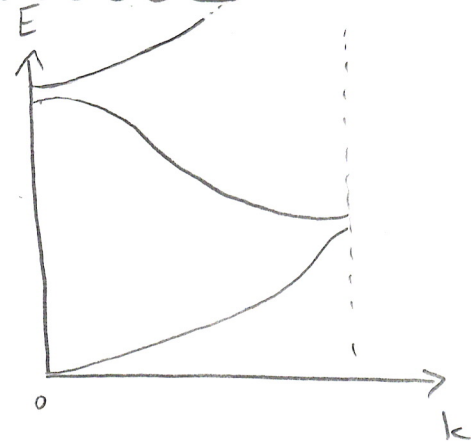
Complex SHO

Hooke's Law

$$V = \frac{1}{2} G x^2$$

~~allllllllllll~~

Electrons



2N
blc
Spin
 $S = \frac{1}{2}$

"Free e⁻ Model"

"Drude Model"

$$KE = \frac{p^2}{2m} \rightarrow \frac{\hbar^2 k^2}{2m}$$

Symmetry Allows Represent

Electrical Conduction

Charges accelerate
in an E field

$$j = \frac{dq}{dt} = -ne \langle v \rangle$$

key is n

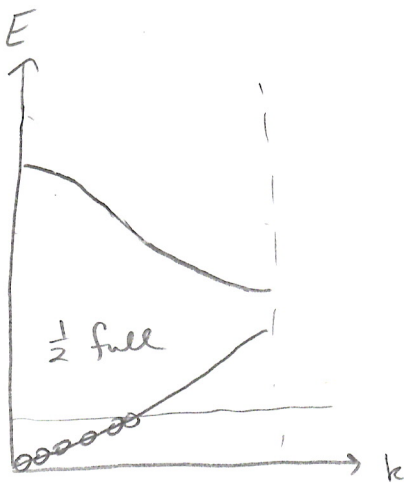
$$\langle v \rangle = -\frac{e\tau}{m} E$$

$$j = \sigma \left(\frac{dV}{dx} \right) = \sigma E$$

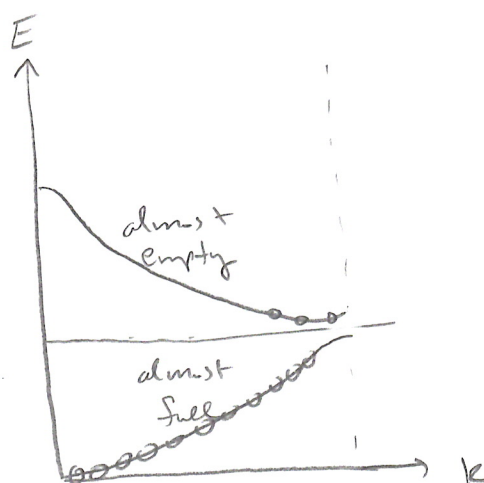
$$\sigma = \frac{ne^2\tau}{m}$$

$$\sigma = \frac{1}{\rho}$$

$$\rho = \frac{RA}{l}$$

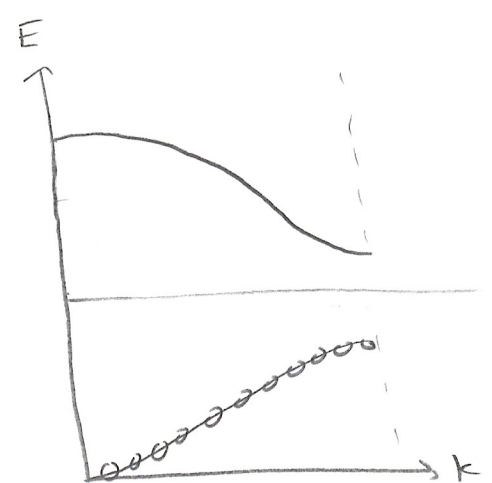


Conductor



Semiconductor

- Intrinsic T
- Extrinsic (Dope)



Insulator

Key Point: Symmetry of \vec{k}

- not full \rightarrow destabilize
- full \rightarrow net zero motion

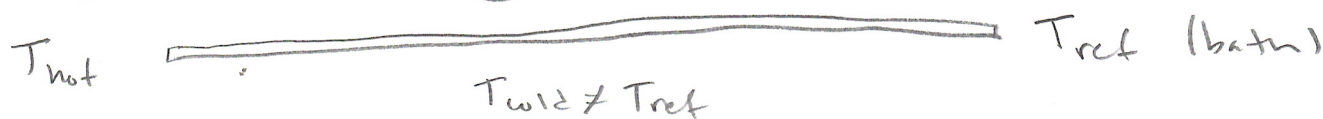
$\Rightarrow n = \# e^-$ in not filled band

Thermal Conduction

$$j = \frac{dQ}{dt} = -\kappa \frac{dT}{dx}$$

- e^- + phonon can carry E $\langle v \rangle$
 - e^- more free \Rightarrow carry more E
- blc e^- dominate !!!
- $\Rightarrow e^-$ + thermal conduct
Similar

Thermal \wedge Electrical "Thermocouple"



Difference in $\langle v \rangle \rightarrow e^-$ build up

$$V = \frac{e \cdot e}{r} \text{ repulsive Coulomb Potential}$$

\Rightarrow opposing E field $\Rightarrow \Delta V$ measure

Summary:

- Bonding intuition \rightarrow detailed quantum bands
- nonmetallic bands play role in semiconductors
- Quantum Atomic \vee Electronic Rules
- Springs or Free e^- qualitatively describe material